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|---|---------------------------------------|--|---------------|
| <b>Module code</b>  | SP-4308                               |  |               |
| <b>Module Title</b>   | Nuclear and Particle Physics          |  |               |
| <b>Degree/Diploma</b>   | Bachelor of Science (Applied Physics) |  |               |
| <b>Type of Module</b>   | Major Option                          |  |               |
| <b>Modular Credits</b>  | 4                                     | <b>Total student Workload</b>  | 10 hours/week |
|   |                                       | <b>Contact hours</b>   | 4 hours/week  |
| <b>Prerequisite</b>   | None                                  |  |               |
| <b>Anti-requisite</b>   | None                                  |  |               |
| <b>Aims</b>   |                                       |  |               |
| The module is designed for students to understand the physics principles underpinning nuclear and particle physics. |                                       |  |               |
| <b>Learning Outcomes</b>  |                                       |  |               |
| <i>On successful completion of this module, a student will be expected to be able to:</i>                           |                                       |  |               |
| Lower order :   | 20%                                   | - describe the patterns of nuclear masses and sizes using simple models and identify the basic constituents of matter and the fundamental forces between them  |               |
| Middle order :  | 50%                                   | - apply calculations involving the energy released by important nuclear decays and reactions<br>- analyse various types of nuclear decay processes using quantitative calculations on radioactivity<br>- apply conservation laws to identify the forces responsible for particular reactions<br>- apply Feynman diagrams to represent elementary processes |               |
| Higher order:   | 30%                                   | - interpret the results of analyses, and make an appropriate report for an effective communication<br>- present case studies or current issues or specific topics individually or collaboratively<br>- work co-operatively in a team   |               |
| <b>Module Contents</b>  |                                       |  |               |
| Nuclear Physics:  |                                       |  |               |
| - Rutherford Scattering, properties of nuclei- Mass, size, charge, magnetic moment                                  |                                       |  |               |
| - Nuclear stability, binding energy and nuclear forces  |                                       |  |               |
| - Nuclear models, The shell model and liquid-drop model, Radioactivity- half-life estimation                        |                                       |  |               |
| - Decay processes, Alpha, Beta & Gamma Decay  |                                       |  |               |
| - Natural Radioactivity- carbon dating, radiation dosage  |                                       |  |               |
| Particle Physics:   |                                       |  |               |
| - Basic properties of cosmic rays, particle accelerators and detectors  |                                       |  |               |
| - The four forces, the quest for unification and links with cosmology   |                                       |  |               |
| - The Standard Model, fermions and their gauge bosons;  |                                       |  |               |
| - Leptons and the electroweak force   |                                       |  |               |
| - The Higgs mechanism and Higgs boson, The strong force, Quarks and gluons  |                                       |  |               |
| <b>Assessment</b>   | Formative assessment                  | In-class questions and feedback  |               |
|   | Summative assessment                  | Examination: 40%<br>Coursework: 60%<br>- 2 work-based problems (20%)<br>- 1 group project (10%)<br>- 1 written assignment (10%)<br>- 1 oral presentation (10%)<br>- 1 class test (10%)   |               |